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'OBI' IS DEFAULT SEARCH FIELD FOR 'HCAPLUS' FILE

=> D STAT QUE L24 L1 ST

Structure attributes must be viewed using STN Express query preparation. L5 45087 SEA FILE=REGISTRY SSS FUL L1

L10 STR

Structure attributes must be viewed using STN Express query preparation. L12 15 SEA FILE=REGISTRY SUB=L5 SSS FUL L10 L14 STR

Page 3 of 44

12 SEA FILE=HCAPLUS SPE=ON ABB=ON PLU=ON (L20 OR L21 OR L22 OR L23) AND (L18 OR L19)

=> D IBIB ED ABS HITSTR L24 1-12

1.24

L24 ANSWER 1 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2009:883577 HCAPLUS Full-text

DOCUMENT NUMBER: 151:276671

TITLE: Homeotropic Alignment of a Discotic Liquid Crystal

Induced by a Sacrificial Layer

AUTHOR(S): Pouzet, Eric; De Cupere, Vinciane; Heintz, Christophe; Andreasen, Jens W.; Breiby, Dag W.;

Nielsen, Martin M.; Viville, Pascal; Lazzaroni, Roberto; Gbabode, Gabin; Geerts, Yves H.

CORPORATE SOURCE: Laboratoire de chimie des polymeres, Universite Libre

de Bruxelles, Brussels, B-1050, Belg.

SOURCE: Journal of Physical Chemistry C (2009), 113(32).

Journal of Physical Chemistry C (2009), 113(32), 14398-14406

CODEN: JPCCCK; ISSN: 1932-7447

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

ED Entered STN: 23 Jul 2009
A convenient method to induce face-on orientation of an alkoxyphtalocyanine discotic mesogen is described. The alignment is imposed by the confinement of the discotic thin films with a top sacrificial polymer layer that is easily removed by washing with a selective solvent, after thermal annealing. Thin films were characterized by optical and atomic force microscopy, UV-visible absorption spectroscopy, and grazing incidence wide angle x-ray scattering. The data converge in showing the central role of the sacrificial layer in promoting alignment with the planar mole. orienting parallel to the substrate in an essentially homeotropic arrangement over large lateral length scales and the persistence of this desirable alignment after removal of the layer.

803724-14-3

CN

RL: PRP (Properties)

(homeotropic alignment of alkoxyphtalocyanine derivative discotic liquid crystal induced by poly(vinylphenol) sacrificial layer)

RN 803724-14-3 HCAPLUS

29H,31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

REFERENCE COUNT: 58 THERE ARE 58 CITED REFERENCES AVAILABLE FOR THIS RECORD, ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 2 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2009:351507 HCAPLUS Full-text DOCUMENT NUMBER: 150:435417

TITLE: Miscibility between Differently Shaped Mesogens:

Structural and Morphological Study of a

Phthalocvanine-Pervlene Binary System

Zucchi, Gael; Viville, Pascal; Donnio, Bertrand; Vlad,

Alexandru; Melinte, Sorin; Mondeshki, Mihail; Graf,

Robert; Spiess, Hans Wolfgang; Geerts, Yves H.

; Lazzaroni, Roberto

CORPORATE SOURCE: Laboratoire de Chimie des Polymeres, Universite Libre

de Bruxelles, Brussels, B-1050, Belg. SOURCE:

Journal of Physical Chemistry B (2009), 113(16), 5448-5457

CODEN: JPCBFK; ISSN: 1520-6106

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

Entered STN: 24 Mar 2009 ED

AUTHOR(S):

AB The thermotropic, structural, and morphol. properties of blends of a disk-like liquid crystalline phthalocyanine derivative and a lath-shaped perylenetetracarboxidiimide mesogen derivative were studied by combining DSC, thermal polarized optical microscopy, x-ray diffraction, solid-state NMR, and atomic force microscopy. The two compds, are fully miscible for blends containing at least 60 mol % of the disk-like mol. In such composition range, the homogeneous blends form a columnar hexagonal (Colh) mesophase for which the thermal stability is enhanced compared to that of the corresponding mesophase of the pure phthalocyanine. The miscible blends self-align homeotropically between two glass slides. For blends containing between 55 and 40 mol % of the disk-shaped mol., the two components are fully miscible at high temperature but the perylene derivative forms a sep. crystalline phase when the temperature is decreased. Phase separation is systematically observed in blends containing <40 mol % of the discotic mol. In this case, the resulting Colh mesophase is less stabilized compared to the blends containing a larger amount of the phthalocyanine derivative These phaseseparated blends do not show any homeotropic alignment. AFM studies confirm the formation of a single columnar morphol. in the phthalocyanine-rich blends, consistent with the full miscibility between the two compds. Solid-state NMR measurements on the mixed phase show the influence of the presence of the perylene mols. on the mol. dynamics of the mols.; remarkably, the presence of the host mols, improves the local order parameter in the phthalocyanine columnar phase.

803724-14-3 1142843-82-0 1142843-86-4 1142843-87-5

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(structural and morphol. study of miscibility, thermal stability and liquid crystal properties of differently shaped phthalocyanine-perylene binary system)

803724-14-3 HCAPLUS RN

CN 29H, 31H-Phthalocyanine, C, C, C, 2-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

RN 1142843-82-0 HCAPLUS

CN Anthra[2,1,9-def:6,5,10-d'e'f']diisoquinoline-1,3,8,10(2H,9H)-tetrone,
2,9-di-(9Z)-9-octadecen-1-yl-, compd. with
C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]-29H,31H-phthalocyanine (1:3) (CA
INDEX NAME)

CM 1

CRN 1017242-09-9

CMF C60 H78 N2 O4

Double bond geometry as shown.

PAGE 1-B

CM 2

CRN 803724-14-3

CMF C128 H210 N8 O4 CCI IDS

RN 1142843-86-4 HCAPLUS

CN Anthra[2,1,9-def:6,5,10-d'e'f']diisoquinoline-1,3,8,10(2H,9H)-tetrone,
2,9-di-(9Z)-9-octadecen-1-yl-, compd. with
C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]-29H,31H-phthalocyanine (1:1) (CA
INDEX NAME)

CM 1

CRN 1017242-09-9

CMF C60 H78 N2 O4

Double bond geometry as shown.

PAGE 1-B

CM 2

CRN 803724-14-3 CMF C128 H210 N8 O4

CCI IDS

RN 1142843-87-5 HCAPLUS

CN Anthra[2,1,9-def:6,5,10-d'e'f']diisoquinoline-1,3,8,10(2H,9H)-tetrone,
2,9-di-(9Z)-9-octadecen-1-yl-, compd. with
C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]-29H,31H-phthalocyanine (3:1) (CA
INDEX NAME)

CM 1

CRN 1017242-09-9

CMF C60 H78 N2 O4

Double bond geometry as shown.



CM 2

CRN 803724-14-3 CMF C128 H210 N8 O4

CCI IDS

REFERENCE COUNT: 36 THERE ARE 36 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 3 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2008:1117198 HCAPLUS Full-text
DOCUMENT NUMBER: 149:365323

TITLE: Femtosecond Charge Transfer in Assemblies of Discotic

Liquid Crystals

AUTHOR(S): de Jong, Michel P.; Osikowicz, Wojciech; Sorensen,

Stacey L.; Sergeyev, Sergey; Geerts, Yves N.

; Salaneck, William R.

CORPORATE SOURCE: Department of Physics, Chemistry and Biology, IFM, Linkoping University, Linkoping, 58183, Swed.

SOURCE: Journal of Physical Chemistry C (2008), 112(40),

15784-15790

CODEN: JPCCCK; ISSN: 1932-7447

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal

LANGUAGE: English

ED Entered STN: 18 Sep 2008

AB The electronic coupling strength within columns of discotic liquid crystals was studied using core-level resonant photoemission spectroscopy. Coexisting well-ordered and disordered regions are identified in thin films of tetra-alkoxy-substituted phthalocyanines with the aid of near edge x-ray absorption fine structure and photoelectron spectroscopies. These different regions were used to derive a lower limit for the intermol. charge transfer bandwidth

within the framework of the core-hole clock principle. Average charge transfer times on the order of a few femtoseconds, i.e., significantly faster than the C(ls) core-hole lifetime, which indicates a surprisingly strong electronic coupling between the phthalocyanine units as compared to what is expected from the charge transport characteristics of this material were found.

803724-14-3

RL: PRP (Properties) (femtosecond charge transfer in assemblies of discotic liquid crystals)

803724-14-3 HCAPLUS RN

29H,31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]- (CA CN INDEX NAME)

REFERENCE COUNT: 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 4 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2008:1015941 HCAPLUS Full-text

DOCUMENT NUMBER: 149:367625

TITLE: Monolayer Control of Discotic Liquid Crystal by Electromigration of Dewetted Lavers in Thin Film

Devices

Calo, Annalisa; Stoliar, Pablo; Cavallini, AUTHOR(S): Massimiliano; Sergevev, Sergev; Geerts, Yves

M.; Biscarini, Fabio

CORPORATE SOURCE: Institute for the Study of Nanostructured Materials,

CNR, Bologna, 40129, Italy Journal of the American Chemical Society (2008),

130(36), 11953-11958

CODEN: JACSAT; ISSN: 0002-7863

PUBLISHER: American Chemical Society

Journal DOCUMENT TYPE:

LANGUAGE: English ED Entered STN: 22 Aug 2008

SOURCE:

AB Ultrathin films of a semiconductive discotic liquid crystal, viz. phthalocyanines, can be organized to form a conductive channel tens of microns long between Au electrodes with thickness control over a single monolayer. The authors' approach exploits the electromigration of the isotropic phase formed starting from the pretransitional region of the columnar-isotropic phase transition. Dewetted isotropic material accumulates to the neg. electrode by applying a longitudinal elec. field of .apprx.1 V/um. Dewetting

and electromigration expose an ultrathin film, a few monolayers thick, exhibiting columnar liquid crystal order. The layers of this ultrathin film melt progressively above TC and can be individually exfoliated by electromigration, starting from the ninth down to the 1st monolayer. The anal. of the current flowing through the junction as a function of the temperature, together with the comparative imaging of the evolution of morphol., yields a detailed picture of the changes in the dimensionality of the conductive phthalocyanine film and allows the authors to extract the behavior of the order parameter. The phenomenon of electromigration opens interesting questions on the technol. control of individual monolayers on device patterns.

IT 870088-23-6 870088-24-7 870088-25-8

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(monolayer control of discotic liquid crystal by electromigration of dewetted layers in thin film devices)

RN 870088-23-6 HCAPLUS

CN

29H,31H-Phthalocyanine, 2,9,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

Me— (CH2) 9

Me— (CH2) 11— CH2— O

Me— (CH2) 9— Me

Me— (CH2) 9

Me— (CH2) 9

Me— (CH2) 9— Me

Me— (CH2) 11— CH2— O

Me— (CH2) 9— Me

PAGE 1-B

- (CH2)11-Me

__ (CH2)11_Me

RN 870088-24-7 HCAPLUS

CN 29H,31H-Phthalocyanine, 2,9,17,24-tetrakis[(2-decyltetradecyl)oxy]- (CA

INDEX NAME)

- RN 870088-25-8 HCAPLUS
- CN 29H, 31H-Phthalocyanine, 2,10,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

- (CH2)11-Me

- (CH2)11-Me

REFERENCE COUNT: 37 THERE ARE 37 CITED

THERE ARE 37 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 5 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2008:741181 HCAPLUS Full-text DOCUMENT NUMBER: 149:359203

TITLE:

Microscopic morphology of thin films of phthalocyanine/perylene blends for organic solar cell

devices

AUTHOR(S): Vlad, Alexandru; Serban, Dana A.; Viville, Pascal; de Cupere, Vinciane; Zucchi, Gael; Melinte,

Sorin: Bayot, Vincent: Lazzaroni, Roberto:

Geerts, Yves

DICE, Universite catholique de Louvain, Louvain-la CORPORATE SOURCE:

Neuve, Bela.

SOURCE: Materials Research Society Symposium Proceedings (2005), 871E(Organic Thin-Film Electronics), No pp.

given, Paper #: 19.43

CODEN: MRSPDH; ISSN: 0272-9172

URL: http://www.mrs.org/s_mrs/bin.asp?CID=2734&DID=149

234&DOC=FILE.PDF PUBLISHER:

Materials Research Society DOCUMENT TYPE: Journal; (online computer file)

LANGUAGE: English ED Entered STN: 20 Jun 2008

AB We report on the microstructure of

2(3)-9(10)-16(17)-23(24)-tetra(2-decyltetradecyloxy)-

phthalocyanine/pervleneolevlamine (PcH2/PTCDI) blends. Thin films, to be used as active layers in organic photovoltaic cells, were prepared by spin coating and spin casting of dilute toluene solns, on indium tin oxide (ITO) substrates. The morphol. of the thin films has been studied using Tapping Mode (TM) atomic force microscopy (AFM), whereas SEM was used to reveal the various top electrode morphologies, inherent to the different film processing

conditions. 1056028-09-1

> RL: TEM (Technical or engineered material use); USES (Uses) (microscopic morphol. of thin films of phthalocyanine/perylene blends for organic solar cell devices)

1056028-09-1 HCAPLUS RN

CN 29H, 31H-Phthalocyanine, 2,9-bis[(1-decyltetradecyl)oxy]-16,23-bis[(2decyltetradecyl)oxy]- (CA INDEX NAME)

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 6 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2007:866756 HCAPLUS Full-text

ACCESSION NUMBER: 2007:866756 HCAPLUS Full-text
DOCUMENT NUMBER: 147:437332

TITLE: Uniaxial Alignment of Nanoconfined Columnar Mesophases

AUTHOR(S): Mouthuy, Pierre-Olivier; Melinte, Sorin; Geerts,

Yves H.; Jonas, Alain M.

CORPORATE SOURCE: Cermin, Universite Catholique de Louvain,

Louvain-la-Neuve, 1348, Belg.

SOURCE: Nano Letters (2007), 7(9), 2627-2632 CODEN: NALEFD; ISSN: 1530-6984

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

ED Entered STN: 09 Aug 2007

AB By confining discotic phthalocyanines in a network of crisscrossed nanogrooves, the authors obtain a uniaxial alignment of the columnar mesophase. The alignment process is based on the anisotropy of interface tension between the mesophase and the nanogrooves' walls. Preferential mesophase alignment results from this nonhomogeneity combined with the anisotropy of the network cell dimensions. A simple model is proposed to explain the explain the explain the explain the such lobservations.

II 870088-23-6 870088-24-7 870088-25-8

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PROC (Process)

(uniaxial alignment of tetrakis[(decyltetradecyl)oxy]phthalocyanine nanoconfined columnar mesophases on PMMA masks on silicon oxidized wafers using anisotropy of interface tension between mesophases and nanocrooves' walls)

RN 870088-23-6 HCAPLUS

CN 29H, 31H-Phthalocyanine, 2,9,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

Me— (CH2) 9

Me— (CH2) 11— CH2— O

N

N

N

N

N

N

N

N

(CH2) 9— Me

(CH2) 9— Me

(CH2) 9— Me

(CH2) 9— Me

PAGE 1-B

- RN 870088-24-7 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,9,17,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

- RN 870088-25-8 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,10,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

- (CH2)11-Me

- (CH2)11-Me

OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD

(2 CITINGS)

REFERENCE COUNT: 18 THERE ARE 18 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 7 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2006:1219525 HCAPLUS Full-text

DOCUMENT NUMBER: 147:129820

TITLE: Charge transport properties of a metal-free

phthalocyanine discotic liquid crystal

AUTHOR(S): Deibel, C.; Janssen, D.; Heremans, P.; De Cupere,

V.; Geerts, Y.; Benkhedir, M. L.;

Adriaenssens, G. J.

CORPORATE SOURCE: IMEC, Louvain, 3001, Belg.

SOURCE: Organic Electronics (2006), 7(6), 495-499

CODEN: OERLAU; ISSN: 1566-1199

PUBLISHER: Elsevier B.V.

DOCUMENT TYPE: Journal LANGUAGE: English

ED Entered STN: 21 Nov 2006

AB Discotic liquid crystals can self-align to form one-dimensional semiconducting wires, many tens of microns long. In this letter, we describe the preparation of semiconducting films where the stacking direction of the disk-like mols. is perpendicular to the substrate surface. We present measurements of the charge carrier mobility, applying temperature-dependent time-of-flight transient photocond. space-charge limited current measurements, and field-effect mobility measurements. We provide exptl. verification of the highly anisotropic nature of semiconducting films of discotic liquid crystals, with

charge carrier mobilities of up to 2.8 + 10-3 cm2/V s. These properties make discotics an interesting choice for applications such as organic

photovoltaics.

IT 803724-14-3

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(charge transport properties of metal-free phthalocyanine discotic liquid crystal)

RN 803724-14-3 HCAPLUS

CN 29H, 31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

OS.CITING REF COUNT: THERE ARE 8 CAPLUS RECORDS THAT CITE THIS RECORD (8 CITINGS)

REFERENCE COUNT: 13 THERE ARE 13 CITED REFERENCES AVAILABLE FOR THIS

RECORD, ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 8 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN

ACCESSION NUMBER: 2006:1205658 HCAPLUS Full-text DOCUMENT NUMBER: 145:513960

TITLE: Method for the manufacturing of homeotropically

aligned layer of discotic liquid crystals

INVENTOR(S): De Cupere, Vinciane; Heintz, Christophe; Geerts, Yves; Tant, Julien

PATENT ASSIGNEE(S): Universite Libre de Bruxelles, Belg.

SOURCE: Eur. Pat. Appl., 14pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.					KIND DATE			APPLICATION NO.							DATE			
EP 1722424					A1	A1 20061115			EP 2005-447108							20050513		
	R:	AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	HU,	ΙE,	
		IS,	ΙT,	LI,	LT,	LU,	MC,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR,	AL,	BA,	
		HR,	LV,	MK,	YU													
ORITY APPLN. INFO.: EP 2005-447108									20050513									

PRIORITY APPLN. INFO.:

ED Entered STN: 16 Nov 2006 The present invention relates to a method for the manufacturing of an

homeotropically aligned layer of discotic liquid crystals comprising the steps of: - depositing a layer of a discotic material on a substrate: - depositing a sacrificial layer on said discotic layer; - inducing the homeotropic alignment of the discotic material of the layer by a magnetic field or by thermal annealing; - removing the sacrificial layer. This allows the liquid crystals to avoid air contact during annealing and allows subsequent deposition of other active organic layers or contacts.

803724-14-3

RL: PEP (Physical, engineering or chemical process); PYP (Physical process); TEM (Technical or engineered material use); USES (Uses)

(method for manufacturing homeotropically aligned layer of discotic liquid crystals)

803724-14-3 HCAPLUS

CN 29H,31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD

(2 CITINGS)
REFERENCE COUNT: 7 THERE ARE 7

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS

RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 9 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2006:478715 HCAPLUS Full-text

DOCUMENT NUMBER: 145:177655

TITLE: Effect of Interfaces on the Alignment of a Discotic

Liquid-Crystalline Phthalocyanine De Cupere, Vinciane; Tant, Julien;

Viville, Pascal; Lazzaroni, Roberto; Osikowicz, Wojciech; Salaneck, William R.; Geerts, Yves

Henri

CORPORATE SOURCE: Laboratory of Polymer Chemistry, Universite Libre de

Bruxelles, Brussels, 1050, Belg. : Langmuir (2006), 22(18), 7798-7806

SOURCE: Langmuir (2006), 22(18), 7798-780 CODEN: LANGD5; ISSN: 0743-7463

PUBLISHER: American Chemical Society
DOCUMENT TYPE: Journal

DOCUMENT TYPE: Journal
LANGUAGE: English
ED Entered STN: 23 May 2006

AB This paper deals with the influence of the nature and number of solid interfaces on the alignment of the columns in a semiconducting discotic liquid crystal. The solid substrates were characterized in terms of their roughness and surface energy. The alignment of the discotic liquid crystal columns on these substrates was determined by optical microscopy under crossed polarizers and by tapping-mode atomic force microscopy. The nature of the substrates has negligible influence on the alignment. The key parameter is the confinement imposed to the film. These surprising observations are explained by the

antagonist alignment role of gas and solid interfaces.

IT 803724-14-3

AUTHOR(S):

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PYP (Physical process); PROC (Process)

(effect of interfaces on alignment of semiconducting discotic

liquid-crystalline phthalocyanine)

RN 803724-14-3 HCAPLUS

CN 29H,31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]- (CA

INDEX NAME)

OS.CITING REF COUNT: 16 THERE ARE 16 CAPLUS RECORDS THAT CITE THIS

RECORD (16 CITINGS)

REFERENCE COUNT: 58 THERE ARE 58 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 10 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2006:80962 HCAPLUS Full-text

DOCUMENT NUMBER: 144:321878

TITLE: Liquid crystalline metal-free phthalocyanines designed for charge and exciton transport. [Erratum to document

cited in CA144:014471]

AUTHOR(S): Tant, Julien; Geerts, Yves Henri;

Lehmann, Matthias; De Cupere, Vinciane;

Zucchi, Gaeel; Laursen, Bo Wegge; Bjornholm, Thomas; Lemaur, Vincent; Marcq, Valerie; Burquel, Anick; Hennebicq, Emmanuelle; Gardebien, Fabrice; Viville, Pascal; Beljonne, David; Lazzaroni, Roberto; Cornil,

Jerome

CORPORATE SOURCE: Laboratoire de Chemie des Polymeres, Universite Libre

de Bruxelles, Brussels, B-1050, Belg.

Journal of Physical Chemistry B (2006), 110(7), 449

CODEN: JPCBFK; ISSN: 1520-6106

PUBLISHER: American Chemical Society

DOCUMENT TYPE: Journal LANGUAGE: English

SOURCE:

ED Entered STN: 29 Jan 2006

AB On page 20319, Table 1 lists incorrect cell parameters; the correct values are given. On page 20319, right column, lines 24 and 25, the halo values were corrected from "4.6 and 5.1 Å" to "4.9 and 5.1 Å". On page 20319, right column, lines 34 through 37, the π-π stacking distance is corrected from "3.5 Å" to "3.4-3.5 Å". On page 20319, right column, lines 51 and 52, the disk diams. are corrected from "1a (30.4 Å) and for 1b (31.8 Å)" to "1a (30.0 Å) and for 1b (32.2 Å)". On page 20320, right column, lines 8 and 9, the exptl. values are corrected from "30.8 and 32.1 Å" to "30.7 and 32.5 Å". On page 20323, Reference 45 is corrected from "Ivanov, D. A. Personal communication" to by "Gearba, R. I.; Bondar, A. I.; Goderis, B.; Brasa, W.; Ivanov, D. A. Chem Mater. 2005, 17, 2825-2832.". The online supporting information is also

IT 870088-18-9P 870088-19-0P 870088-20-3P

870088-21-4P 870088-22-5P 870088-23-6P 870088-24-7P 870088-25-8P RL: PEP (Physical, engineering or chemical process); PRP (Properties); PYP (Physical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process) (preparation, liquid crystal properties and photophysics and mol. dynamics isomeric mixture containing (Erratum)) 870088-18-9 HCAPLUS 29H, 31H-Phthalocyanine, 2,9,16,23-tetrakis[(2-octyldodecyl)oxy]- (CA

Me (CH2) 7

Me (CH2) 9

Me (CH2) 7

Me (CH2) 9

Me (CH2) 7

Me (CH2) 9

PAGE 1-B

— Ме

of

RN

CN

INDEX NAME)

RN 870088-19-0 HCAPLUS

CN 29H, 31H-Phthalocyanine, 2,9,16,24-tetrakis[(2-octyldodecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

RN 870088-20-3 HCAPLUS

CN 29H,31H-Phthalocyanine, 2,9,17,24-tetrakis[(2-octyldodecyl)oxy]- (CA INDEX NAME)

RN 870088-21-4 HCAPLUS

CN 29H, 31H-Phthalocyanine, 2,10,16,24-tetrakis[(2-octyldodecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

PAGE 1-A

- RN 870088-22-5 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,9,16,23-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

$$\begin{array}{c} \text{Me-} (\text{CH}_2) \, 9 \\ \text{Me-} (\text{CH}_2) \, 11 - \text{CH-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{CH-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{CH-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} \text{CH}_2 - 0 \\ \\ \text{Me-} (\text{CH}_2) \, 11 - \text{Me-} (\text{CH}_2) \\$$

- RN 870088-23-6 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,9,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

__ (CH2)11-Me

- RN 870088-24-7 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,9,17,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

- RN 870088-25-8 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,10,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

- (CH2)11-Me

- (CH2)11-Me

OS.CITING REF COUNT: 2 THERE ARE 2 CAPLUS RECORDS THAT CITE THIS RECORD (2 CITINGS)

L24 ANSWER 11 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NOMBER: 2005:1085985 HCAPLUS Full-text DOCUMENT NUMBER: 14:14471

TITLE: Liquid Crystalline Metal-Free Phthalocyanines Designed

for Charge and Exciton Transport
AUTHOR(S): Tant, Julien; Geerts, Yves Henri;

Lehmann, Matthias; De Cupere, Vinciane;

Zucchi, Gaeel; Laursen, Bo Wegge; Bjornholm, Thomas; Lemaur, Vincent; Marcq, Valerie; Burquel, Anick; Hennebicq, Emmanuelle; Gardebien, Fabrice; Viville, Pascal; Beljonne, David; Lazzaroni, Roberto; Cornil,

Jerome

CORPORATE SOURCE: Laboratoire de Chimie des Polymeres, Universite Libre

de Bruxelles, Brussels, B-1050, Belg.

SOURCE: Journal of Physical Chemistry B (2005), 109(43),

20315-20323

CODEN: JPCBFK; ISSN: 1520-6106

PUBLISHER: American Chemical Society
DOCUMENT TYPE: Journal

LANGUAGE: English
ED Entered STN: 11 Oct 2005

AB A joint theor, and exptl. study of the electronic and structural properties of liquid crystalline metal-free phthalocyanines bearing a strong potential for charge and exciton transport was performed. The synthesis of such compds. was triggered by quantum chemical calcns. showing that: (i) hole transport is favored in metal-free phthalocyanines by their extremely low reorganization energy (0.045 eV) and large electronic splittings; and (ii) the efficiency of energy transfer along the 1-dimensional discotic stacks is weakly affected by rotational disorder due to the two-dimensional character of the mols. The authors synthesized two metal-free phthalocyanines with different branched aliphatic chains on the gram scale to allow for a full characterization of their solid-state properties. The two compds. self-organize in liquid crystalline mesophases, as evidenced by optical microscopy, DSC, x-ray powder diffraction, and mol. dynamics simulations. They exhibit a columnar rectangular mesophase at room temperature and a columnar hexagonal mesophase at elevated temperature

IT 870088-18-9P 870088-19-0P 870088-20-3P 870088-21-4P 870088-22-5P 870088-23-6P

870088-24-7P 870088-25-8P

RL: PEP (Physical, engineering or chemical process); PRP (Properties); PYP (Physical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(preparation, liquid crystal properties and photophysics and mol. dynamics

isomeric mixture containing)

RN 870088-18-9 HCAPLUS

of

CN 29H,31H-Phthalocyanine, 2,9,16,23-tetrakis[(2-octyldodecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

- RN 870088-19-0 HCAPLUS CN 29H,31H-Phthalocyanine
 - N 29H,31H-Phthalocyanine, 2,9,16,24-tetrakis[(2-octyldodecyl)oxy]- (CA INDEX NAME)

PAGE 1-B

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___Ме

RN 870088-20-3 HCAPLUS CN 29H,31H-Phthalocyanine, 2,9,17,24-tetrakis[(2-octyldodecyl)oxy]- (CA INDEX NAME)

- RN 870088-21-4 HCAPLUS

PAGE 1-B

— Ме

— Ме

- RN 870088-22-5 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,9,16,23-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

PAGE 1-A

$$\begin{array}{c} \text{Me} - (\text{CH}_2) \, 9 \\ \text{Me} - (\text{CH}_2) \, 11 - \text{CH} - \text{CH}_2 - \text{O} \\ \text{Me} - (\text{CH}_2) \, 11 - \text{CH} - \text{CH}_2 - \text{O} \\ \text{Me} - (\text{CH}_2) \, 9 \\ \text{Me} - (\text{CH}_2) \, 11 - \text{CH} - \text{CH}_2 - \text{O} \\ \text{Me} - (\text{CH}_2) \, 11 - \text{CH}_2 - \text{O} \\ \text{Me} - (\text{CH}_2) \, 11 - \text{Me} \\ \text{Me} - (\text{CH}_2)$$

PAGE 1-B

- (CH2)11-Me

- RN 870088-23-6 HCAPLUS
- CN 29H, 31H-Phthalocyanine, 2,9,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

PAGE 1-A

PAGE 1-B

- RN 870088-24-7 HCAPLUS
- CN 29H,31H-Phthalocyanine, 2,9,17,24-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

$$\begin{array}{c} \text{Me}-\text{(CH2)}9\\ \text{Me}-\text{(CH2)}11-\text{CH}-\text{CH2}-\text{O}\\ \text{N}\\ \text{N}\\ \text{N}\\ \text{N}\\ \text{N}\\ \text{N}\\ \text{N}\\ \text{Me}-\text{(CH2)}9-\text{Me}\\ \\ \text{Me}-\text{(CH2)}9\\ \text{Me}\\ \text{Me}-\text{(CH2)}9\\ \text{Me}\\ \text{Me}-\text{(CH2)}9\\ \text{Me}\\ \text{Me}-\text{(CH2)}11-\text{Me}\\ \end{array}$$

RN 870088-25-8 HCAPLUS

CN 29H,31H-Phthalocyanine, 2,10,16,24-tetrakis[(2-decyltetradecyl)oxy]- (CA

INDEX NAME)

PAGE 1-B

- (CH2)11-Me

- (CH2)11-Me

OS.CITING REF COUNT: 25 THERE ARE 25 CAPLUS RECORDS THAT CITE THIS

RECORD (25 CITINGS)

REFERENCE COUNT: 56 THERE ARE 56 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L24 ANSWER 12 OF 12 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2005:810842 HCAPLUS Full-text

DOCUMENT NUMBER: 143:220437

TITLE: Phthalocyanine derivative layer in electronic

multilayer devices and method for the manufacturing

thereof

INVENTOR(S): De Cupere, Vinciane; Tant, Julien;

Geerts, Yves

PATENT ASSIGNEE(S): Universite Libre de Bruxelles, Belg.

SOURCE: Eur. Pat. Appl., 9 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent

LANGUAGE: English

FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 1564826	A1	20050817	EP 2004-447032	20040210
D. AT BE	CH DE DK	ES FR GR	GR IT I.I I.II NI.	SE MC DT

```
IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, HU, SK
     CA 2555309
                                20050818 CA 2005-2555309 20050118
                          A1
     WO 2005076383
                                 20050818 WO 2005-EP556
                          A2
     WO 2005076383
                          A3
                                 20051124
         W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH,
             CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD,
             GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC,
             LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI,
             NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY,
             TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW, SM
         RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM,
             AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, MC, NL, PL, PT,
             RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML,
             MR, NE, SN, TD, TG
                          A2
                                 20061025
                                            EP 2005-701089
         R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT,
             IE, SI, LT, FI, RO, CY, TR, BG, CZ, EE, HU, PL, SK, IS
     US 20070225491
                               20070927
                                            US 2007-588865
                         A1
                                                                     20070608
                                                                  A 20040210
PRIORITY APPLN. INFO.:
                                             EP 2004-447032
                                             WO 2005-EP556
                                                                 W 20050118
```

OTHER SOURCE(S): MARPAT 143:220437

ED Entered STN: 18 Aug 2005

AB The present invention relates to an electronic device including at least one organic semi-conducting layer comprising a homeotropically organized phthalocyanine derivative sandwiched between at least two substrate layers. The electronic device can be used in photovoltaic cells, organic light emitting diodes and sensors.

803724-14-3D, derivs.

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process); USES (Uses) (phthalocyanine derivative layer in electronic multilayer devices and

method for the manufacturing thereof)

RN 803724-14-3 HCAPLUS

CN 29H, 31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

REFERENCE COUNT:

3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD, ALL CITATIONS AVAILABLE IN THE RE FORMAT

Structure Search

=> D STAT QUE L18 L1 STR

Structure attributes must be viewed using STN Express query preparation. L5 $$45087\ \rm SEA\ FILE=REGISTRY\ SSS\ FUL\ L1$ L10 $$\rm STR$

Structure attributes must be viewed using STN Express query preparation. L12 15 SEA FILE-REGISTRY SUB-L5 SSS FUL L10 L18 6 SEA FILE-RCAPLUS SPE-ON ABB-ON PLU-ON L12

=> S L18 NOT L24 L39 1 L18 NOT L24 => D IBIB ED ABS HITSTR 1

L39 ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2009 ACS on STN ACCESSION NUMBER: 2006:1066869 HCAPLUS Full-text

DOCUMENT NUMBER: 145:440958

TITLE: Lubricant composition

INVENTOR(S): Kawata, Ken

PATENT ASSIGNEE(S): Fuji Photo Film Co., Ltd., Japan SOURCE:

PCT Int. Appl., 98pp. CODEN: PIXXD2

DOCUMENT TYPE: Patent

LANGUAGE: Japanese FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

PATENT NO.					KIND DATE				APPL								
WO 2006106856																	
	W:	ΑE,	AG,	AL,	AM,	AT,	AU,	ΑZ,	BA,	BB,	BG,	BR,	BW,	BY,	BZ,	CA,	CH,
		CN,	CO,	CR,	CU,	CZ,	DE,	DK,	DM,	DZ,	EC,	EE,	EG,	ES,	FI,	GB,	GD,
		GE,	GH,	GM,	HR,	HU,	ID,	IL,	IN,	IS,	KE,	KG,	KM,	KN,	KP,	KR,	KZ,
		LC,	LK,	LR,	LS,	LT,	LU,	LV,	LY,	MA,	MD,	MG,	MK,	MN,	MW,	MX,	MZ,
		NA,	NG,	NI,	NO,	NZ,	OM,	PG,	PH,	PL,	PT,	RO,	RU,	SC,	SD,	SE,	SG,
		SK,	SL,	SM,	SY,	TJ,	TM,	TN,	TR,	TT,	TZ,	UA,	UG,	US,	UZ,	VC,	VN,
		YU,	ZA,	ZM,	ZW												
	RW:	AT,	BE,	BG,	CH,	CY,	CZ,	DE,	DK,	EE,	ES,	FI,	FR,	GB,	GR,	HU,	IE,
		IS,	IT,	LT,	LU,	LV,	MC,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR,	BF,	ВJ,
		CF,	CG,	CI,	CM,	GA,	GN,	GQ,	GW,	ML,	MR,	NE,	SN,	TD,	TG,	BW,	GH,
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		KG,	KZ,	MD,	RU,	TJ,	TM										
JP	2006	3072	01		A		2006	1109		JP 2	006-	9362	0		2	0060	330
JP	2006	3072	02		A		2006	1109		JP 2	006-	9362	20060330				
EP	1876	220			A1 20080109					EP 2	006-	7306	20060330				
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		IS,	IT,	LI,	LT,	LU,	LV,	MC,	NL,	PL,	PT,	RO,	SE,	SI,	SK,	TR	
US	2009	0143	262		A1		2009	0604		US 2	007-	8874	51		2	0070	928
KR	2007	1166	72		A		2007	1210		KR 2	007-	7249	89		2	0071	029
IORIT	Y APP	LN.	INFO	. :						JP 2	005-	9891	7		A 2	0050	330
										WO 2	006-	JP30	6718		W 2	0060	330
HER S	OURCE	(S):			MAR	PAT	145:	4409	58								
En:	tered	STN	: 1	3 Oc	t 20	06											
Di	sclos	sed i	is a	lubi	icar	it c	ompos	itic	n c	ontai	ning	, a r	olym	ner v	hich	has	a mes

OT

- ED
- AB structure in a main chain or a side chain. The polymer added as the viscosity index-improving agent also improves shear stability and lubricating characteristics of the lubricant oil.
- 912821-61-5 912822-21-0 912823-09-7
 - RL: MOA (Modifier or additive use); USES (Uses)
- (viscosity index improving polymer additive for lubricant oils)
- 912821-61-5 HCAPLUS
- Poly[[9,10,16,17,23,24-hexakis[(2-hexyldecyl)oxy]-29H,31H-phthalocyanine-2,3-diyl-KN29, KN30, KN31, KN32] (1,12-dioxo-2,5,8,11tetraoxadodecane-1,12-diyl) (SP-4-2)-copper complex] (9CI) (CA INDEX NAME)

^{*} STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT *

- * STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY AVAILABLE VIA OFFLINE PRINT *
- * STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY AVAILABLE VIA OFFLINE PRINT *

- * STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY AVAILABLE VIA OFFLINE PRINT *
- * STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY AVAILABLE VIA OFFLINE PRINT *

^{*} STRUCTURE DIAGRAM TOO LARGE FOR DISPLAY - AVAILABLE VIA OFFLINE PRINT * RN 912822-21-0 HCAPLUS

CN Copper, [2-[2-[3,9,10,16,17,23,24-heptakis](2-hexyldecyl)oxy]-29H,31H-

phthalocyanin-2-y1-kN29,kN30,kN31,kN32]oxy]ethoxy]
ethoxy[ethy1 2-methy1-2-propenoato(2-)]-, (SP-4-2)-, homopolymer (9CI)
(CA INDEX NAME)

CM 1

CRN 912822-20-9 CMF C154 H256 Cu N8 O12

CCI CCS

PAGE 1-A

Me-

PAGE 2-A

Me___

RN 912823-09-7 HCAPLUS

CN Copper, [9,10,16,17,23,24-hexakis[(2-hexyldecyl)oxy]-29H,31H-phthalocyanine-2,3-dicarboxylato(4-)
kN29,kN30,kN31,kN32]-, dihydrogen, (SP-4-2)-,
polymer with 2,2'-[1,2-ethanediylbis(oxy)]bis[ethanol] (9CI) (CA INDEX NAME)

CM 1

CRN 912823-08-6

CMF C130 H206 Cu N8 O10 . 2 H

CCI CCS

PAGE 1-A

Me-

PAGE 2-A

Me-

●2 H+

PAGE 2-B

CM 2

CRN 112-27-6 CMF C6 H14 O4

HO-CH2-CH2-O-CH2-CH2-O-CH2-CH2-OH

REFERENCE COUNT:

7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT => D STAT QUE L19 L1 STR

Structure attributes must be viewed using STN Express query preparation. L5 $$45087\ \text{SEA}$$ FILE-REGISTRY SSS FUL L1 L14 \$5TR\$

Structure attributes must be viewed using STN Express query preparation. L16 5 SEA FILE-REGISTRY SUB-L5 SES FUL L14 L19 9 SEA FILE-REGIDUS SPE-ON ABB-ON PLU-ON L16

=> S L19 NOT L18, L24 L40 2 L19 NOT (L18 OR L24)

=> D IBIB ED ABS HITSTR 1-2

L40 ANSWER 1 OF 2 HCAPLUS COPYRIGHT 2009 ACS On STN ACCESSION NUMBER: 2008:1350942 HCAPLUS Full-text DOCUMENT NUMBER: 150:110464

TITLE: Comparative structural study of thin films of a

columnar liquid crystal aligned by mechanical shearing

and zone casting

Anokhin, Denis V.; Rosenthal, Martin; Makowski, AUTHOR(S):

Tomasz; Tracz, Adam; Bras, Wim; Kvashnina, Kristina;

Ivanov, Dimitri A.

CORPORATE SOURCE: CNRS UPR 9069, Institut de Chimie des Surfaces et Interfaces, Mulhouse, F-68057, Fr.

Thin Solid Films (2008), 517(2), 982-985

SOURCE:

CODEN: THSFAP: ISSN: 0040-6090

PUBLISHER: Elsevier B.V. DOCUMENT TYPE: Journal LANGUAGE: English

ED Entered STN: 11 Nov 2008

Zone casting is a promising method for fabrication of open highly oriented crystalline and liquid crystalline (LC) films for various applications in (opto)electronics. The authors have performed a comparative structural anal. of mech. sheared and zone-cast films of a model columnar LC. Grazing incidence x-ray diffraction and UV-vis spectroscopy show that, contrary to the mech. sheared films, the columns in the zone-cast films are aligned perpendicular to the casting direction. In the films, two LC domains with [20] or [11] reciprocal space vectors perpendicular to the substrate plane are observed This can be explained by a small lattice mismatch allowing epitaxial growth of the LC domains on each other.

ΙT 803724-14-3

AΒ

RL: PEP (Physical, engineering or chemical process): PRP (Properties): PROC (Process)

(comparative structural study of thin films of columnar liquid crystal aligned by mech. shearing and zone casting)

RN 803724-14-3 HCAPLUS

CN 29H, 31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxyl- (CA INDEX NAME)

REFERENCE COUNT: 15 THERE ARE 15 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

L40 ANSWER 2 OF 2 HCAPLUS COPYRIGHT 2009 ACS on STN 2004:1054303 HCAPLUS Full-text ACCESSION NUMBER:

DOCUMENT NUMBER: 142:45544

TITLE. A composition for photon-energy up-conversion INVENTOR(S): Miteva, Tzenka; Nelles, Gabriele; Yasuda, Akio; Balouchev, Stanislav; Keivanidis, Panagiotis; Lupton,

John

PATENT ASSIGNEE(S): Sony International Europe G.m.b.H., Germany;

Max-Planck-Gesellschaft zur Foerderung der

Wissenschaften e.V. SOURCE: Eur. Pat. Appl., 30 pp.

CODEN: EPXXDW

DOCUMENT TYPE: Patent

LANGUAGE: English FAMILY ACC. NUM. COUNT: 1

PATENT INFORMATION:

	PATENT NO.						KIND DATE			APPLICATION NO.						DATE		
	EP 1484379			A1 20041208				2003-	1253	20030602								
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			IE,	SI,	LT,	LV,	FI,	RO,	MK,	CY,	AL,	TR,	BG,	CZ,	EE,	HU,	SK	
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	AU	2004	2024	13		A1		2004	1216		AU 2	2004-	2024	13		2	0040	531
	AU	2004	2024	13		B2		2009	0730									
	KR	2004	1035	00		A		2004	1208		KR 2	2004-	3953	9		2	0040	601
	JP	2005	0498	24		A		2005	0224		JP 2	2004-	1647	83		2	0040	602
PRIO	RIT	Y APP	LN.	INFO	. :						EP 2	2003-	1253	6		A 2	0030	602
	Title 4		O.Mar		0 D-	- 00	0.4											

ED Entered STN: 09 Dec 2004

AB A composition for photon energy up-conversion is described comprising at least two components, wherein a first component is capable of absorbing energy at a first wavelength region $\lambda 1$, which first component acts as a sensitizer in the composition, and wherein a second component is capable of emitting energy at a second wavelength region $\lambda 2$, which second component acts as an emissive component in the composition, wherein $\lambda 2 \leq \lambda 1$, and wherein, upon absorption of energy by the first component at the first wavelength region $\lambda 1$, the emissive component emits energy at the second wavelength region $\lambda 2$, characterized in that the first component and/or the second component is an organic compound A photon-energy upconversion system comprising a substrate and the described composition is also described.

803724-13-2 803724-14-3

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(sensitizer; composition for photon-energy up-conversion and devices of using the same)

RN 803724-13-2 HCAPLUS

CN 29H, 31H-Phthalocyanine, C,C,C,2-tetrakis[(2-octyldodecyl)oxy]- (9CI) (CA INDEX NAME)

$$^{4}\left[\begin{array}{c} _{\text{CH}_{2}}\text{CH}_{2}\text{-O-D1} \\ _{\text{Me}_\text{(CH}_{2})}\text{7}_\text{CH}_\text{(CH}_{2})}\text{9}_\text{Me} \end{array} \right]$$

RN 803724-14-3 HCAPLUS

CN 29H, 31H-Phthalocyanine, C,C,C,2-tetrakis[(2-decyltetradecyl)oxy]- (CA INDEX NAME)

OS.CITING REF COUNT: 10 THERE ARE 10 CAPLUS RECORDS THAT CITE THIS RECORD (10 CITINGS)

REFERENCE COUNT: 7 THERE ARE 7 CITED REFERENCES AVAILABLE FOR THIS RECORD. ALL CITATIONS AVAILABLE IN THE RE FORMAT

Search History

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L1
             STRUCTURE UPLOADED
L2
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   FILE 'HCAPLUS' ENTERED AT 14:41:28 ON 16 SEP 2009
L3
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    FILE 'REGISTRY' ENTERED AT 14:41:59 ON 16 SEP 2009
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L.4
              803724-14-3/BI)
        45087 SEA SSS FUL L1
L6
           50 SEA SPE=ON ABB=ON PLU=ON L5 AND L2
1.7
            1 SEA SPE=ON ABB=ON PLU=ON L5 AND L4
L8
              STRUCTURE UPLOADED
L9
          50 SEA SUB=L5 SSS SAM L8
L10
              STRUCTURE UPLOADED
1.11
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L12
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L13
            O SEA SPE=ON ABB=ON PLU=ON L12 AND L4
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L14
L15
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L16
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L17
            1 SEA SPE=ON ABB=ON PLU=ON L16 AND L4
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L19
            9 SEA SPE=ON ABB=ON PLU=ON L16
           19 SEA SPE=ON ABB=ON PLU=ON DE CUPERE V?/AU
L20
L21
           14 SEA SPE=ON ABB=ON PLU=ON TANT J?/AU
L22
          94 SEA SPE=ON ABB=ON PLU=ON GEERTS Y?/AU
L23
            O SEA SPE=ON ABB=ON PLU=ON DECUPERE V?/AU
           12 SEA SPE=ON ABB=ON PLU=ON (L20 OR L21 OR L22 OR L23) AND
L24
               (L18 OR L19)
    FILE 'WPIX' ENTERED AT 15:18:17 ON 16 SEP 2009
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L26
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L27
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L31
            0 SEA SSS SAM L14
L32
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L33
         2381 SEA SSS FUL L1
L34
          153 SEA SUB=L33 SSS FUL L10
L35
            0 SEA SUB=L33 SSS FUL L14
   FILE 'MARPAT' ENTERED AT 15:31:55 ON 16 SEP 2009
L36 STRUCTURE UPLOADED
L37
              STRUCTURE UPLOADED
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1.40		2 SEA	SPE=ON	ABB=ON	PLU=ON	L19 NOT (L18	OR L24)